

MY TWENTY YEARS WITH POLYAROMATIC COMPOUNDS

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Abstract:

We have developed synthetic methods for the preparation of new classes of anticancer agents starting from polyaromatic compounds. In this perspective, we describe some of our research on polyaromatic compounds that have found attention to the scientific community.

Introduction:

During the selection of the chemistry honors students at the Bejoy Narayan College, Itachuna, Hooghly, West Bengal, India, the interviewer faculty members had asked me to explain them the reasons I would pursue a career as a chemist. My response was that I want to become a research scientist. The interviewer started smiling. They told me whether I know the difficulty in becoming a researcher and the sacrifices I have to make in my life. Although I was not sure about the exact difficulties and challenges, I knew it would be a very hard choice. I answered to my two interviewer and others who were present in the room I would do well in research and you can follow me in the future. I had a dream from my childhood that I would conduct research on cancer areas that are not explored well and/or controversial. This feeling helped me to undertake unusual research: synthesis and biological evaluation of polyaromatic compounds as anticancer agents. The two professors (Professor P. K. Ghosh and Professor J. Nath) at the Bejoy Narayan College I mentioned above are really become very pleased to know my whole career path. Their blessings are priceless and I will continue to adore them because they never neglected my feelings what I said to them at the age of 17.

Significance:

DNA is the principal target for numerous small molecules. Organic molecules that bind to DNA constitute a classic class of molecules in anticancer therapy.¹ However, DNA binding is not only the criterion to develop anticancer molecules. An interaction with DNA is often necessary to maintain reasonable cytotoxic effects. DNA intercalators are characterized by the insertion of planar aromatic or heteroaromatic systems in between DNA base pairs.² Molecules like anthracene, acridines and ellipticines have good DNA intercalator properties due to their topoisomerase inhibition activities.³ Hydrogen bonding and stacking are important in the stabilization of the compound-DNA complex. Polycyclic aromatic compounds and their derivatives have received significant attention from chemists, pharmacologists and

biologists.⁴ However, an insignificant amount of research is focused on the use of compounds related to polycyclic aromatic compounds as anticancer molecules.⁵ Bair *et al.* demonstrated a correlation between antitumor activity and the nature of the polyaromatic compounds.⁶ Despite their efforts, a conclusive correlation between the ability of these molecules to bind to DNA and their cytotoxic activity could not be determined. This study culminated in the development of a few benzylic aminopropanediols from a structure-activity relationship work. These amino propanediols interacted with DNA by intercalation and they acted as a topoisomerase II inhibitors. Clinical studies on the lead molecules were not successful due to CNS and neurological toxicity. Naphthalimides, amonafide and mitonafide were selected for clinical trials based on their sufficient activity due to intercalation, but their clinical results were weak. Naphthalene ring was changed to anthracene in order to determine the role of the chromophore. The resulting product, azonafide proved to be effective in increasing cellular cytotoxicity over naphthalene-related compounds.⁷ Azonafides molecules which demonstrated similar mechanistic characteristics when compared with common DNA intercalators, do not localize in the cell nucleus. This was an interesting finding. Other crucial differences with existing DNA intercalators, for example clinically useful mitoxantrone^{8,9}, were the lack of inhibition of topoisomerase II enzymes at equivalent molar cytotoxic concentrations.

Based upon a series of scientific literature myths that polyaromatic compounds are carcinogenic and mutagenic, a scientist at the beginning of his/her career may not pursue research on them with the hope that novel anticancer agents can be derived from these types of molecules. These two properties are highly contradictory: The first well-established science describes polyaromatic compounds as cancer promoter. The second issue whether they can be used for the development of molecules that can inhibit the growth of cancer cells or even cure the disease. It is obvious that the second issue is very complicated. It is a highly questionable approach to develop polyaromatic compounds as anticancer molecules. Being an aspiring scientist at the University of Texas M. D. Anderson Cancer Center, Houston, USA, I undertook the challenge in 1995 with the background information that mustards, carbazoles and anthracenes are active anticancer agents, yet many of them are highly toxic.¹⁰ The concept was further strengthened that scientist mainly focus interaction of DNA and RNA interaction with organic drug candidates. Highly hydrophobic and lipophilic polyaromatic compounds can also demonstrate interaction with various lipids present in the cell membrane as well as with cell nucleus.^{11, 12} One can think something special, but realization of the special ideas need implementations. Based upon some literature background, we undertook a program of synthesis and evaluating of novel polyaromatic compounds as anticancer agents in 1995. This work has given us publications, patents, grants as well as recognition.

Results:

An analysis of chrysene derivatives *in vitro* against a panel of human tumor cell lines was performed.¹² During the course of the investigation, a number of new synthetic methods for the preparation of polyaromatic amines were developed. Synthesis of this type of compound required different types of amines (6-aminochrysene, 2-aminochrysene, dibenzofluorene, fluorene, 9-aminophenanthrene, 3-amino-(9,10)-dihydrophenanthrene and 9-aminophenanthridine) as the starting compounds. A facile aromatic nitration reaction of an aromatic compound by bismuth nitrate using solid support was developed.¹³ This versatile method was applied for the preparation of several aromatic nitro compounds in an excellent yield. In

contrast to other methods, bismuth nitrate-induced nitration of aromatic substrates was highly efficient, regioselective and environmentally friendly. New methods for the reduction of polycyclic aromatic nitro compounds with samarium and indium to the corresponding amines were developed in good yield¹⁴. The amine was then converted to several diamides and diamines. The compounds were tested against several cancer cell lines *in vitro* using cisplatin as the control. Some of the chrysene derivatives demonstrated significant cytotoxic activity against a broad spectrum of tumor cell lines. Some general conclusions can be drawn from these results: (1) In general, the 2- and 6-chrysene derivatives demonstrated equal cytotoxicity; (2) Amide compounds with a terminal piperazine heterocyclic ring were more effective than those with a piperidine terminal ring; (3) the amino derivatives demonstrated enhancement of cytotoxic activity. However, two compounds were found to be extremely active against the seven cell lines tested (IC₅₀ values 0.5-2.5 μ M). They were more potent than cisplatin *in vitro*. The *in vitro* activity was close to adriamycin. The antitumor activity was then evaluated against a number of human tumors implanted in nude mice. Both solid tumors and hematologic malignant tumors were tested. The lead compound demonstrated good activity against both a human ovarian tumor (SKOV-3) and a human colon adenocarcinoma (HT-29). This compound was also submitted to NCI for evaluation in their Developmental Therapeutics Program and it passed the initial screen for relative activity against the human tumor cell line. Complete inhibition of cell growth was produced against four of six human colon cell lines and against three of eight melanoma cell lines. When the COMPARE algorithm was used to try to discern the potential mechanism of action of this molecule, NCI reported that the mechanism of cytotoxicity of this compound did not match that of standard cytotoxic compounds. Thus, the lead molecule may have a unique mechanism of action that would require further research to elucidate. Using the results of the three 60-cell panel runs the drug evaluation group of the NCI inserted the data in the program "COMPARE" aimed at detecting mechanisms of action of lead compounds.

Interestingly, the active chrysene compounds had not demonstrated mutagenic properties. Importantly, apoptotic pathways through caspase-3 activation played a crucial role in inducing cell death in human leukemia Jurkat T cells by several of these compounds that contain chrysene as their aromatic ring system.¹² No effect was demonstrated in a normal non-transformed line of human natural killer cells. In addition, these compounds were relatively non-toxic to mice and non-mutagenic when tested in bacterial systems. These results provide evidence for the potential chemotherapeutic use of chrysene-related compounds. Based on the results, it was also expected that phenanthrene and dihydrophenanthrene compounds may not demonstrate any carcinogenic and mutagenic properties. This conclusion was proved to be true. These compounds demonstrated activity against human colon and leukemia cancer cells by arresting cell division at the G₂/M cell cycle checkpoint. This target is considered to be of considerable therapeutic interest.

The high degree of caspase-3 activation, PARP cleavage and DNA-damage demonstrated by the TUNEL assay as a result of exposure of lead compounds clearly suggested apoptotic pathways involved in resulting cell death. One of the most striking findings was the absence of effect of compound against a non-transformed, natural killer cell line. This further supported the probability that the effect of this compound against leukemic Jurkat T cells targets some aspect of tumor cell physiology. There was no significantly detectable increase in caspase-3 activation at

very lower concentration of the amide types of compound. This finding suggests that alternate pathways exist for inducing cell damage, such as, direct interaction with the cells' surfaces for amides.

This study required a few specific molecules in order to determine the effects of structures. Specifically, we developed new oxidation method¹⁵, dimerization of ester¹⁶, reductive cyclization reaction¹⁷ for the preparation of new compounds of polyaromatic nature. Functionalization of some of these molecules proved difficult to achieve an identical goal as described above. The work on polyaromatic compounds prompted us to prepare these new chemical entities and consequently, new method development was necessary. Our synthetic chemistry expertise was highly necessary to prepare new polyaromatic structures with unique features. Some of the compounds resulted in from this study showed important properties for further elaboration not only in synthetic chemistry, but also to study their biochemical mechanism of actions.¹⁸

Conclusion:

In contrast to literature, our research group demonstrated the preparation of several polyaromatic compounds that exhibited reasonable *in vitro* and *in vivo* activity against a number of tumor cell lines with unique mechanism of actions. Notable the lead compounds had no effects on normal cells. Results described herein were very useful to study the synthesis, biological evaluation and mechanism of action of new polyaromatic compounds. This field of research was regarded as risky in 1995. However, systematic studies with available knowledge were very helpful in the identification of novel polyaromatic molecules that showed anticancer activities. In addition, several new reactions and reagents were developed during the course of this investigation. We are pursuing this subject extensively.

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